

GENERAL BATTERY CORP -- ELECTROLYTE (ACID) BATTERY FLUID -- 6140-00-059-3528

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Product Identification
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Product ID:ELECTROLYTE (ACID) BATTERY FLUID

MSDS Date:09/01/1985

FSC:6140

NIIN:00-059-3528

MSDS Number: BPBRL

=== Responsible Party ===

Company Name:GENERAL BATTERY CORP

Address:645 PENN STREET

Box:14205

City:READING

State:PA

ZIP:19612-4205

Country:US

Info Phone Num:215-378-0527

Emergen

cy Phone Num:215-378-0527

Preparer's Name:J. A. BITLER

CAGE:06810

=== Contractor Identification ===

Company Name:EXIDE CORP.-GENERAL BATTERY CORP

Address:645 PENN STREET

Box:City:READING

State:PA

ZIP:19601

Country:US

Phone:215-378-0527/800-424-9300(CHEMTREC)

CAGE:08163

Company Name:GENERAL BATTERY CORPORATION

Box:1262

City:READING

State:PA

ZIP:19603

Country:US

Phone:215 378-0500

CAGE:06810

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Composition/Information on Ingredients
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Ingred Name:SULFURIC ACID (SARA III)

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CAS:7664-93-9
RTECS #:WS5600000
Fraction by Wt: 32-40%
Other REC Limits:NONE RECOMMENDED
OSHA PEL:1 MG/M3
ACGIH TLV:1 MG/M3; 9293
EPA Rpt Qty:1000 LBS
DOT Rpt Qty:1000 LBS

Ingred Name:WATER
CAS:7732-18-5
RTECS #:ZC0110000
Fraction by Wt: 60-68%
Other REC Limits:NONE RECOMMENDED

===== Hazards Identification =====

LD50 LC50 Mixture:ORAL LD50 (RAT) IS 2140 MG/KG
Routes of Entry: Inhalation:YES Skin:YES Ingestion:YES
Reports of Carcinogenicity:NTP:NO IARC:
NO OSHA:NO

Health Hazards Acute and Chronic:EYES: SEVERE BURNS, CORNEA DAMAGE &
BLINDNESS, SKIN: SEVERE IRRITATION, BURNS & ULCERATION. INHALATION:
VAPORS/ MIST MAY CAUSE SEVERE RESPIRATORY IRRITATION AT LEVELS OF
5MG/M3. INFLAMMATION OF BRONCHIAL MEMBRANES MAY OCCUR AT
12-35MG/M3. INGESTION: MAY CAUSE SEVERE BURNS & ULCERATION OF
MOUTH/THROAT/ESOPHAGUS/STOMACH.

Explanation of Carcinogenicity:THIS PRODUCT IS NOT LISTED BY IARC, NTP
OR OSHA AS A CARCINOGEN, MUTAGEN, TERATOGEN OR NEUROTOXIN.

Effects of Overexposure:MAY CAUSE SEVERE BURNS AND ULCERATION
ACCOMPANIED BY STINGING SENSATION. CHRONIC EXPOSURES ARE KNOWN TO
CAUSE EROSION OF THE TEETH ACCOMPANIED BY DISCOLORATION,
INFLAMMATION OF NOSE, THROAT AND BRONCHIAL TUBES. MAY CAUSE COUGH,
INCREASED RESPIRATORY RATE AND IMPAIRMENT OF VENTILATORY CAPACITY.

Medical Cond Aggravated by Exposure:EXPOSURE TO MIST MAY CAUSE LUNG
DAMAGE AND AGGRAVATE PULMONARY CONDITIONS.

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First Aid Measures =====

First Aid: EYES: IMMEDIATELY FLUSH WITH PLENTY OF WATER FOR AT LEAST 15 MINUTES. GET MEDICAL ATTENTION. SKIN: IMMEDIATELY FLUSH WITH PLENTY OF WATER WHILE REMOVING CONTAMINATED CLOTHING. INHALATION: REMOVE TO FRESH AIR. IF NOT BREATHING, GIVE ARTIFICIAL RESPIRATION. IF BREATHING IS DIFFICULT, GIVE OXYGEN. INGESTION: DO NOT INDUCE VOMITING. IF CONSCIOUS, GIVE WATER OR MILK TO DRINK. GET MEDICAL ATTENTION.

===== Fire Fighting Measures =====

Flash Point: NON-FLAMMABLE
Autoignition Temp: Autoignition Temp Text: NONE
Extinguishing Media: NONCOMBUSTIBLE. USE EXTINGUISHING MEDIA APPROPRIATE FOR SURROUNDING FIRE CONDITIONS.
Fire Fighting Procedures: WEAR SELF CONTAINED BREATHING APPARATUS AND FULL COVER SULFURIC ACID RESISTANT PROTECTIVE CLOTHING.
Unusual Fire/Explosion Hazard: WATER APPLIED TO SULFURIC ACID GENERATES HEAT AND CAUSES SPATTERING. REACTS VIOLENTLY WITH METALS; NITRATES; AND ORGANIC MATERIALS, RELEASING HYDROGEN GAS.

===== Accidental Release Measures =====

Spill Release Procedures: PROVIDE VENTILATION & CONFINE SPILL. DO NOT ALLOW RUNOFF TO SEWER. CLEAN-UP PERSONNEL SHOULD USE RESPIRATORY AND LIQUID CONTACT PROTECTION. KEEP COMBUSTIBLES AWAY FROM SPILLED MATERIAL. ABSORB IN VERMICULITE, DRY SAND OR EARTH & PLACE INTO CONTAINER.
Neutralizing Agent: LIME, SODA ASH, SODIUM BICARBONATE, ETC.

===== Handling and Storage =====

Handling and Storage Precautions: STORE IN WELL VENTILATED, COOL, DRY AREA. SEPARATE FROM INCOMPATIBLE MATERIALS. KEEP CONTAINER TIGHTLY CLOSED AND PROTECTED FROM PHYSICAL DAMAGE.
Other Precautions: STORE AND HANDLE ONLY IN AREAS WITH UNLIMITED WATER SUPPLY AND WHERE SPILLS CAN BE CONTROLLED. AVOID SPILLS, LEAKS OR DAMAGE TO CONTAINERS. WHEN DILUTING, ALWAYS POUR ACID SLOWLY INTO WATER.

===== Exposure Controls/Personal Protection =====

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Respiratory Protection:NONE REQUIRED UNDER NORMAL CONDITIONS. WHEN CONCENTRATIONS OF SULFURIC ACID IN AN AREA ARE KNOWN TO EXCEED PEL, USE NIOSH-APPROVED, FITTED, FULL FACE RESPIRATOR WITH ACID GAS CANISTER OR HEPA OR ANY SUPPLIED AIR RESPIRATOR OR SCBA.

Ventilation:LOCAL EXHAUST: PREFERRED. MECHANICAL: ACCEPTABLE IF CONCENTRATION STAYS BELOW PEL. USE ACID & EXPLOSION PROOF EQUIPMENT.

Protective Gloves:ACID RESISTANT RUBBER OR PLASTIC GLOVES

Eye Protection

:CHEMICAL SPLASH GOGGLES W/FACE SHIELD

Other Protective Equipment:WEAR ACID-RESISTANT APRON. UNDER SEVERE EXPOSURE OR EMERGENCY CONDITIONS, WEAR ACID-RESISTANT CLOTHING AND BOOTS.

Work Hygienic Practices:WASH HANDS AND FACE AFTER HANDLING MATERIAL. LAUNDRY CONTAMINATED CLOTHING BEFORE REUSE. DISCARD CONTAMINATED FOOTWEAR.

Supplemental Safety and Health

WHEREVER SULFURIC ACID IS HANDLED IN CONCENTRATIONS GREATER THAN 1%, PROVIDE EMERGENCY EYEWASH AND SHOWER OR SIMILAR FACILITIES.

WATER

SUPPLY SHOULD BE AT LEAST 0.4 GALLON PER MINUTE FOR EYEWASH AND 3.0 GALLON PER MINUTE FOR BODY FLUSHING, AND SUFFICIENT TO CONTINUE FLUSHING FOR AT LEAST 15 MINUTES.

===== Physical/Chemical Properties =====

HCC:C1

Boiling Pt:B.P. Text:203F,95C

Melt/Freeze Pt:M.P/F.P Text:-35F,-37C

Vapor Pres:10

Vapor Density:>1

Spec Gravity:1.245-1.295

pH:1

Evaporation Rate & Reference: